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UNITED STATES PATENT AND TRADEMARK OFFICE

BEFORE THE BOARD OF PATENT APPEALS AND INTERFERENCES

Ex parte TSUYOSHI KATAYAMA, MASAKAZU OKUMURA, NOBUAKI HATTORI, MAKOTO NAKAJIMA and OSAMU KIMURA

> Application 09/604,763 Technology Center 1600

Oral Argument: 05 December 2007 Decided: 22 January 2008

Before: FRED E. McKELVEY, Senior Administrative Patent Judge, and ADRIENE LEPIANE HANLON and MARK NAGUMO, Administrative Patent Judges.

Opinion for the Board filed by Senior Administrative Patent Judge McKELVEY.

DECISION ON APPEAL

A. Statement of the case

1

3 Tsuyoshi Katayama, Masakazu Okumura, Nobuaki Hattori, Makoto 4Nakajima, and Osamu Kimura (hereafter "Fine Chemical") seek review

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Appeal 2007-4123
Application 09/604,763
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- We have jurisdiction under 35 U.S.C. § 6(b).
- 4 The application on appeal was filed on 26 June 2000.
- Fine Chemical claims benefit of an earlier filing date based on 6Japanese patent application 11-181497, filed 28 June 1999.
- The real party in interest is Nippon Fine Chemical Co., Ltd. (Fine 8Chemical).
- 9 The Examiner rejected claims 17-20, 22-29, and 33 under 35 U.S.C. 10§ 103(a) as being unpatentable over Ansmann and Akrongold. (The reader 11should know that no references to *et al.* are made in this opinion.)
- 12 The Examiner has also rejected claim 21 under 35 U.S.C. § 103(a) 13over Ansmann, Akrongold, and Bernhardt.
- 14 The following specific prior art was relied upon by the Examiner.

13			
16	<u>Name</u>	Patent Number	Issue Date
17	Akrongold	US 3,846,550	05 Nov. 1974
18	Bernhardt	US 4,788,054	29 Nov. 1988
19	Ansmann	US 5,795,978	18 Aug. 1998

21 Ansmann, Akrongold, and Bernhardt are prior art under 35 U.S.C. 228 102(b).

23 We also call attention to:

15

20

26	Harrison and	Kojima are prior art unde	r 35 U.S.C. § 102(b).
25	Kojima	WO98/08888	05 Mar. 1998
24	Harrison	US 2,731,481	17 Jan. 1956

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Appeal 2007-4123
  Application 09/604,763
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        B. Record on appeal
        In deciding this appeal, we have considered only the following
 3
 4documents:
             1. Specification, including original claims.
 5
 6
             2. Final Rejection entered 04 June 2003 (Paper 19)
             3. Appeal Brief filed 06 January 2004
 7
 8
             4. Examiner's Answer entered 06 May 2004
             5. Reply Brief filed 06 July 2004
 9
10
             6. Examiner's transmittal to the Board entered 19 July 2007
             7. PTO bibliographic data sheet for the application on appeal
11
             8. Ansmann
12
             9. Akrongold
13
            10. Bernhardt
14
15
            11. Kojima
            12. Harrison
16
17
            13. Hawley's Condensed Chemical Dictionary, page 583
18(12th ed. 1993)
19
            14. Evidence presented by Fine Chemical:
                    a. Ansmann, U.S. Patent 6,264,961 B1
20
                    b. Ansmann, U.S. Patent 6,235,702 B1
21
22

 c. Ansmann, U.S. Patent 6,033,652

23
                    d. Schrader, U.S. Patent 5,981,452
24

    e. Brunelle, U.S. Patent 5,231,161

25
                    f. Igarashi, U.S. Patent 5,576,408
                    g. Obiols, U.S. Patent 5,880,299
26
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Appeal 2007-4123
Application 09/604,763

1 h. Baumoeller, U.S. Patent 5,888,487

2 i. Ansmann, U.S. Patent 5,939,081

3 j. Wachter, U.S. patent 5,962,663

4 k. 1 International Cosmetics Ingredient Dictionary

5488-489, 521-522 (5th ed. 1993).

6 C. Issues

- 7 Two issues are raised by the Appeal Brief.
- The first issue is whether Fine Chemicals has sustained its burden of 9showing that the Examiner erred in rejecting claims 17-20, 22-29, and 33 on 10appeal as being unpatentable under 35 U.S.C. § 103(a) over Ansmann '978 11and Akrongold.
- The second issue is whether Fine Chemicals has sustained its burden 13of showing that the Examiner erred in rejecting claim 21 on appeal as being 14unpatentable under 35 U.S.C. § 103(a) over Ansmann '978, Akrongold and 15Bernhardt.
- We do not reach the first issue with respect to claim 17 and 22-29. 17Instead, we entered a new ground of rejection of claims 17 and 22-29 as 18being indefinite within the meaning of the second paragraph of 35 U.S.C. 198 112.

20

21 D. Findings of fact

The following findings of fact are believed to be supported by a 23preponderance of the evidence. To the extent that a finding of fact is a 24conclusion of law, it may be treated as such. Additional findings as 25necessary may appear in the Discussion portion of the opinion.

	Appeal 2007-4123
	Application 09/604,763
7	
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The invention

- The invention described in the specification relates to (1) "oil 3materials" comprising (a) an ester of "dimerdiol" with a monocarboxylic 4acid having 4 to 34 carbon atoms or (b) an ester of a dimerdiol with a 5dicarboxylic acid and (2) cosmetics and external agents comprising either 6dimerdiol ester. Specification 1:5-9; 2:15-21.
- 7 The claims on appeal are limited to the cosmetics and external agent 8portion of the invention.
- 9 According to the specification, "[t]he dimerdiols and/or esters thereof 10 for producing the dimerdiol carboxylate and oils comprising the ester of the 11 present invention are known." Specification 2:24-26.
- Fine Chemicals has the following to say about "dimerdiols." 13Specification 3:1 through 4:21 [material in brackets added]:

A dimer acid [which is not a dimer diol] is a known dibasic acid obtainable by an intermolecular polymerization reaction of an unsaturated fatty acid, and the industrial production process thereof is approximately standardized in the art. For example, [1] a dimer acid and/or a lower alcohol ester thereof can be obtained by dimerization of an unsaturated fatty acid having 11 to 22 carbon atoms and/or [2] a lower alcohol ester thereof with a clay catalyst.

An industrially obtainable dimer acid is mainly composed of a dibasic acid having about 36 carbon atoms. It also contains a trimer acid and monomer acid in any amount depending on the degree of purification. In general, those in

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Application 09/604,763
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which the content of a dimer acid is over 70 wt% and those in which the content of a dimer acid has been increased to 90% or more are commercially available. Further, those which oxidation stability has been improved by hydrogenation of double bonds remaining after the dimerization reaction are also commercially available. In the present invention, any of dimer acids thus commercially available ... can be used.

An industrially obtainable dimerdiol contains other component[s], for example, a trimer triol, monoalcohol, and ether compound[s], depending on the degree of purification of a dimer acid and/or a lower alcohol ester thereof used as a raw material. In general, those in which the content of a dimerdiol is over 70% wt% can be used in the present invention, although a high purity dimerdiol, such as a dimerdiol in which its content is over 90 wt%, is preferable.

A dimerdiol produced by hydrogenating a dimer acid obtained by dimerization of an unsaturated fatty acid having 11 to 22 carbon atoms with a clay catalyst usually contains 70 to 100 wt% of a diol component. It is considered that the dimerdiol mainly contains compounds represented by the following structural formula 1 and/or structural formula 2:

Formula 1 is reproduced below:

$$CH_3(CH_2)_m - CR - CH - (CH_2)_n - OH$$

 $CH_3(CH_2)_p - CH - (CH_2)_q - OH$ (1)

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Appeal 2007-4123
  Application 09/604,763
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 1
                    Formula 1 represents dimerdial compounds
 2
 3
              wherein, each of m, n, p and q independently represents an
 4
              integer and m+n+p+q is from 14 to 36:
                    Formula 2 is reproduced below:
 5
               CH3(CH2)x-CH -(CH2)x-OH
                                                                              (2)
               CR4(CH4)+-CH - (CH4)--OH
 6
                    Formula 2 represents dimerdial compounds
 7
             wherein, each of r, s, t and u independently represents an
 8
 9
              integer and r+s+t+u from 18 to 40.
10
        With respect to the monocarboxylic acid, Fine Chemicals tells us that
11
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12the monocarboxylic acid used in the invention is not particularly restricted 13providing it has 4 to 34 carbon atoms, preferably 10 to 32 carbon atoms. 14Specification 4:22-24.

Numerous examples of suitable monocarboxylic acids are set out in 16the specification. Specification 4:24 through 6:14. Examples include 17(1) butanoic acid [having four carbon atoms], (2) oleic acid [having 18 18carbon atoms and one double bond], (3) linoleic acid [having 18 carbon 19atoms and 2 double bonds], (4) linolenic acid [having 18 carbon atoms and 203 double bonds], (5) abietic acid [having 20 carbon atoms, a ring structure 21and 2 double bonds in the ring structure], (6) hydrogenated rosin and 22(7) rosin.

According to *The Merck Index*, page 1422 (12th ed. 1996), rosin is 24made up of about 90% resin acids and 10% neutral matter. Of the resin 25acids about 90% are isomeric with abietic acid ($C_{20}H_{30}O_2$): the other 10% is a

1mixture of dihydroabietic acid $(C_{20}H_{32}O_2)$ and dehydroabietic acid $2(C_{20}H_{38}O_2)$. Abietic acid has the structure:

3 The structure of abjetic acid is shown below:

5

- 6
- Synthesis Example 1 in the specification describes the preparation of a 8dimerdiol from a mixture of methyl oleate [the methyl ester of oleic acid] 9and methyl linoleate [the methyl ester of linoleic acid]. A dimerization 10reaction was said to have been conducted for 6 hours using activated clay to 11thereby obtain an intermediate product. The dimerdiol is obtained by 12hydrogenation of the intermediate product under 250 atmospheres of 13hydrogen at 250 °C in the presence of a catalyst. The dimerdiol is said to 14have an acid value of 0.2 and a hydroxyl value of 196.0. Specification 21:26 15through 22:17.
- Example 1 describes the preparation of a hydrogenated rosin to dimerdiol ester made by reacting hydrogenated rosin with the dimerdiol sprepared in Synthesis Example 1.
- 19 Example 5 describes the preparation of an ester made from the 20dimerdiol PESPOL HP-1000 which is said to be manufactured by Toagosei

	eal 2007-4123 lication 09/604,763
	Ltd. Specification 24:20. According to WO98/08888, published
205 N	March 1998, page 3:7-12:
3	Dimer diol is a saturated aliphatic diol having 36 carbon
4	atoms prepared by a complete hydrogenation of dimer acid.
5	The dimer acid used as a starting material is obtained by
6	dimerization, with heating, of unsaturated fatty acid having 18
7	carbon atoms such as oleic acid or linoleic acid. For example,
8	commercially-available products such as Pespol HP-1000
9	which is sold from Tagosei Co., Ltd., may be used as the dimer
10	diol.
11	Claims on appeal
12	Claims 17-20, 22-29, and 33 are on appeal.
13	Independent claim 29 reads:
14	29. A cosmetic or an external agent comprising a dimerdiol
15	ester of a monocarboxylic acid having 10 to 32 carbon atoms
16	and/or a dimerdiol ester of a dicarboxylic acid.
17	
18	Independent claim 33 reads:
19	33. A cosmetic or an external agent comprising a dimerdiol
20	ester of a monocarboxylic acid selected from the group
21	consisting of:
22	i) linear saturated acids having 4 to 34 carbon atoms,
23	ii) branched fatty acids having 4 to 34 carbon atoms,
24	iii) liner unsaturated acids having 10 to 32 carbon atoms,
25	iv) hydroxy acids having 4 to 34 carbon atoms and

Appeal 2007-4123 Application 09/604,763		
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v) cyclic acids having 4 to 34 carbon atoms, selected from		
2 the group consisting of cyclohexanoic acid, hydrogenated rosin,		
3 rosin, abietic acid, hydrogenated abietic acid, benzoic acid,		
4 p-oxybenzoic acid, p-aminobenzoic acid, salicyclic acid, gallic		
5 acid, pyrrolidonecarboxylic acid and nicotinic acid; and/or a		
6 dimerdiol ester of a dicarboxylic acid, and		
7 wherein said dimerdiol is a dimerdiol produced by		
8 hydrogenating a dimer acid obtained by dimerization of an		
9 unsaturated fatty acid having 11 to 22 carbon atoms.		
Dependent claims 18 and 21 read:		
11 18. The cosmetic or an external agent according to claim 33,		
wherein the dimerdiol ester is of a monocarboxylic acid having		
13 10 to 32 carbon atoms.		
14		
15 21. The cosmetic or an external agent according to claim 18,		
wherein the monocarboxylic acid comprises a rosin or a		
17 hydrogenated rosin.		
18		
19 <u>Ansmann '978</u>		
Ansmann '978 relates to "emulsifiers." Col. 1:12.		
21 The problem said to be addressed by the Ansmann invention is to		
22provide emulsifiers which enable more stable, but at the same time		
23sensorially lighter products to be produced with a reduced quantity of waxes.		
24Col. 1:31-35.		

- The Ansmann emulsifiers may be used as skin-care formulations in a 2variety of cosmetics, including day creams, night creams and body lotions. 3Col. 5:41-43.
- The Ansmann emulsifier comprises at least two essential ingredients 5as well as other optional ingredients. Col. 1:40-47.
- 6 A first ingredient is an alkyl and/or alkenyl oligoglycoside. 7Col. 1:41-42.
- 8 A second ingredient is a fatty alcohol. Col. 1:44-45.
- 9 The fatty alcohols include "technical dimerdiols and trimertriols 10containing 18 to 36 or 18 to 54 carbon atoms which emanate from the 110ligomerization and subsequent hydrogenation of unsaturated fatty acids." 12Col. 3:13-16
- The emulsifiers are suitable for the production of emulsions of the o/w 14[oil in water] type. Col. 4:40-41.
- 15 Suitable oils include esters of linear and/or branched fatty acids with 16polyhydric alcohols (for example dimer diol or trimer diol). Col. 4:40-41.
- 17 The oil can make up 5–99 weight %, preferably 10–75 w%, of the 18non-aqueous part of the emulsion.
- 19 <u>Akrongold</u>
- 20 We find it unnecessary to discuss Akrongold since it does not discuss 21"dimerdiols."
- 22 <u>Bernhardt</u>
- Bernhardt describes "thickeners" (a viscosity modifier) which can be 24added to sunscreen compositions. One thickener is described as being the 25reaction products of rosin and a polyhydric alcohol. Co. 8:56-67.

	1 2007-4123 ation 09/604,763
1	Fine Chemical contention with respect to dimer diol
2	One of Fine Chemical's principal contentions in this appeal is
3(App.	Br. 5):
4	that "dimer diol" and "trimer diol" discussed at column 4,
5	line 49 of Ansmann '978 are different compounds from
6	"dimerdiol" as used in the present invention, because Ansmann
7	'978 defines "dimerdiol and "trimertriol" in its column 3,
8	lines 14-17 as compounds containing 18 to 36 or 18 to 54
9	carbon atoms obtained from the oligomerization and subsequent
0	hydrogenation of unsaturated fatty acids.
1	Fine Chemical goes on to say (App. Br. 5):
2	Moreover, it appears that a "dimer diol" as used in Ansmann
3	['978] is represented by the formula HO—Ar—O—Ar—OH,
4	wherein Ar is ethylene, i.e.:
5	HO — CH_2CH_2 — O — CH_2CH_2 — OH .
6	Ansmann '978 reveals the following at col. 3:14-17:
7	technical dimerdiols and trimertriols containing 18 to 36 or 18
8	to 54 carbon atoms which emanate from oligomerization and
9	subsequent hydrogenation of unsaturated fatty acids.
:0	What is a "dimer dio1"?
:1	(1)
2	An "oligomer" is "[a] polymer molecule consisting of only a few
3monor	ner units (dimer, trimer, tetramer). Hawley's Condensed Chemical
4Dictio	nary, page 583 (12th ed. 1993). A "dimer" is "[a]n oligomer whose

25molecule is composed of two molecules of the same chemical composition."

11d. at 409. A dimer acid is produced by dimerization of unsaturated fatty 2acids at mid-molecule and usually contains 36 carbons. 1d. at 409.

3 Oleic acid has the structural formula (18 carbon atoms):

- A dimer acid of oleic acid results from the oligomerization of two 6molecules of oleic acid (*i.e.*, two oleic acid molecules are bonded together 7through oligomerization and the oligomer has 36 carbon atoms). Possible 8chemical structures are described by Harrison (col. 2:38-48). Subsequent 9hydrogenation eliminates any double bonds and ultimately yields a dimer 10diol having 36 carbon atoms.
- 11 The process is described by Fine Chemicals. Specification, page 2:24 12through page 4:21.
- 13 A similar process is described by Ansmann '978. Col. 3:14-17 14(quoted above).

- 16 As earlier noted, Ansmann '978 in addition to referring to technical 17dimerdiols, also refers to "dimer diols," Col. 4:48.
- 18 According to Fine Chemical, a "dimer diol" is different from a 19technical dimerdiol.
- At least one prior art document cited by Fine Chemical tends to 21confirm the possibility that one skilled in the art would understand that the 22term "dimerdiol" has more than one meaning.
- 23 Brunelle, a patent assigned to The General Electric Company, 24describes esters having the following formula:

1where n is about 1 to about 5. Col. 2:30-40.

- While the formula represents a compound which is an ester (because 3it contains —O—CO— groups), it is also a dialcohol or a diol because it 4contains —OH groups on either end.
- Brunelle reveals that the "bis-esters" are sometimes designated 6hereinafter "monomer diol" (where n=1), "dimer diol" (where n=2), etc. 7Col. 2:46-48.
- 8 Consistent with Brunelle, is International Cosmetic Ingredient 9Dictionary, called to our attention by Fine Chemical, where various esters 10are described.
- 11 For example, PEG-2 distearate (described on page 488) is a bis-ester 12or di-ester of a "diol" having the formula

14The formula represents a dimer diol due to the fact that two (O—CH₂CH₂—) 15groups are present. Fine Chemical tells us that one skilled in the art would 16understand the diol is a "dimer diol."

17 PEG-3 distearate (described on pages 488-489) is a bis-ester or di-18ester of a "diol" (not a triol) having the formula:

20The formula represents a trimer diol due to the fact that three 21(O—CH₂CH₂—) groups are present. Fine Chemical tells us that one skilled 22in the art would understand that the diol is a "trimer diol."

27

Examiner's reliance on Ansmann '978

- The Examiner rejected all claims on appeal as being unpatentable 3under 35 U.S.C. \$ 103 in one fashion or another over Ansmann '978.
- The Examiner found that Ansmann '978 describes an emulsifier suitable for the production of light oil-in-water emulsions that may be used 6in the cosmetic formulations. Final Rejection, page 2.
- The Examiner further found that suitable oils for the emulsions 8described by Ansmann '978 include esters of linear and/or branched fatty 9acids with polyhydric alcohols, for example *dimer diols. Id. See* Ansmann 10'978, col. 4:27-28 to which references has been earlier made.
- The Examiner's finding of a description of a "dimer diol" in Ansmann 12'798 forms the underlying basis for the prior art rejections.

13

14 E. Principles of law

- A claim containing language which is indefinite is not patentable.

 1635 U.S.C. § 112, second paragraph; *General Electric Co. v. Wabash*17Appliance Corp., 304 U.S. 364 (1938); *In re Moore*, 439 F.2d 1232 (CCPA 181971).
- A rejection based on § 103 should not be reached if the claims are 20indefinite within the meaning of 35 U.S.C. § 112. *In re Steele*, 305 F.2d 21859, 863 (CCPA 1962). *See also In re Wilson*, 424 F.2d 1382, 1385 (CCPA 221970).
- Claims are interpreted from the standpoint of a person of ordinary 24skill in the relevant art. *In re American Academy of Science Tech Center*, 25367 F.3d 1359, 1364 (Fed. Cir. 2004).

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Appeal 2007-4123
Application 09/604,763
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In supporting an obviousness rejection, it can be important to identify 2a reason that would have prompted a person of ordinary skill in the relevant 3 field to combine prior art elements to arrive at the claimed invention. *KSR* 4 International Co. v. Teleflex, Inc., 127 S. Ct. 1727, 1741 (2007).

5

6 F. Discussion

- Indefiniteness of claims 17 and 22-29
- 8 During our deliberations in connection with this appeal, we have 9struggled with the meaning of the term "dimerdiol".
- The term "dimerdiol" appears in the record in two lexical forms: (1) 11"dimer diol" (with a space between "dimer" and "diol") and (2) "dimerdiol." 12We attribute no particular scientific significance to the lexical difference.
- 13 Independent claim 29 uses the term "dimerdiol."
- 14 Independent claim 33 also uses the term "dimerdiol" but adds that the 15"dimerdiol is a dimerdiol produced by hydrogenating a dimer acid obtained 16by dimerization of an unsaturated fatty acid having 11 to 22 carbon atoms."
- 17 The "dimerdiol" of claim 29 facially appears to be broader than the 18dimerdiol of claim 33
- 19 The specification describes dimerdiols and indicates that they are 20known. Specification, page 2:24 through page 4:21.
- It turns out from the prior art, that dimerdiols are indeed old and that 22one skilled in the art reasonably could understand that a reference to a 23"dimerdiol" could be a reference to at least two diols which are different, 24e.g., those described by Brunelle and those described by WO98/08888. The 25"dimer diols" (with a space between "dimer" and "diol") described by 26WO98/08888 appear to be one embodiment of the "dimerdiols" described in

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Application 09/604,763
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1the Specification. Notwithstanding the examples provided in the 2specification, we cannot read limitations from the specification into the 3claims. *Constant v. Advanced Micro-Devices, Inc.*, 848 F.2d 1560, 1571 4(Fed. Cir. 1988) ("particular embodiments and examples appearing in the 5specification will not generally be read into the claims.")

- Whether a claim is indefinite can be the subject of expensive patent 7litigation. See, e.g., IPXl Holdings, L.L.C. v. Amazon.com, Inc., 430 F.3d 81377 (Fed. Cir. 2005); Datamize, LLC v. Plumtree Software, Inc., 417 F.3d 91342 (Fed. Cir. 2005); Genentech Inc. v. Wellcome Foundation Ltd., 29 F.3d 101555 (Fed. Cir. 1994).
- When indefinite claim language becomes apparent during patent 11 12examination, the time to clear up the indefiniteness is before a patent issues. 13Cf. Graham v. John Deere Co., 383 U.S. 1, 18 (1966) (to await litigation is 14—for all practical purposes—to debilitate the patent system); In re Zletz, 839 15F.2d 319, 321-22 (Fed. Cir. 1989) ("[d]uring patent examination the pending 16claims must be interpreted as broadly as their terms reasonably allow. . . . 17The reason is simply that during patent prosecution when claims can be 18amended, ambiguities should be recognized, scope and breadth of language 19explored, and clarification imposed. . . . An essential purpose of patent 20examination is to fashion claims that are precise, clear, correct, and 21unambiguous. Only in this way can uncertainties of claim scope be 22removed, as much as possible, during the administrative process."). In the 23case before us, a simple amendment might overcome any indefiniteness and. 24should a patent issue to Fine Chemical, spare patentees, accused infringers. 25and Federal courts from the need to guess about, battle over, and resolve the

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Appeal 2007-4123
Application 09/604,763
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1meaning of term claims. *See, e.g., Saunders Group, Inc. v. Comfortrac, Inc.*, 2492 F.3d 1326, 1335-36 (Fed. Cir. 2007), where both the Federal Circuit and 3the district court struggled with the meaning of claim language.

- 4 The dimerdiol of claim 29 has no carbon atom limitation.
- 5 The dimerdiol of claim 33 has a carbon atom limitation by virtue of 6the "product by process" language.
- One possibility for an amendment may be to define "dimerdiol" in 8claim 29 as the product obtained by hydrogenating a dimer acid obtained by 9dimerization of an unsaturated fatty acid having 11 to 22 carbon atoms. 10Specification, page 3:28 through page 4:3.
- 11 Still another possibility would be an amendment to claim 29 to define 12the "dimerdiol" as comprising mainly compounds represented by formula 1 13or formula 2. Specification, page 4:4-21.
- Both possibilities would eliminate from the scope of claim 29 the 15Brunelle dimer diols and other prior art dimer diols called to our attention by 16Fine Chemical in its brief on appeal. It is our understanding that Fine 17Chemical does not intend the claimed term "dimerdiol" to cover the dimer 18diols described by Brunelle.

Obviousness

Claim 33 defines the dimerdiol as a dimerdiol produced by 21hydrogenating a dimer acid obtained by dimerization of an unsaturated fatty 22acid having 11 to 22 carbon atoms. We do not believe that the evidence 23relied upon by the Examiner would permit us to find by a preponderance of 24the evidence that the dimer diol described by Ansmann (col. 4, line 49) is a 25dimerdiol which is the same as a dimerdiol produced by hydrogenating a

1dimer acid obtained by dimerization of an unsaturated fatty acid having 211 to 22 carbon atoms. The dimer diol described by Ansmann '978 may or 3may not be a dimerdiol produced by hydrogenating a dimer acid obtained by 4dimerization of an unsaturated fatty acid having 11 to 22 carbon atoms. 5Possibilities and speculation are not sufficient to establish a fact by a 6preponderance of the evidence. *Central State Hospital v. Wiggers*, 230 Va. 7157, 159 (1985). For example, inherency cannot be established by a 8preponderance of the evidence based on evidence that a certain thing *may* 9result from a given set of circumstances. *Rapoport v. Dement*, 254 F.3d 101053 (Fed. Cir. 2001). *See also In re Hughes*, 345 F.2d 184 (CCPA 1965) 11(if a reference is subject to two interpretations, then it is ambiguous and will 12not support an anticipation rejection).

- Since a preponderance of the evidence does not support a finding that 14Ansmann '978 describes a dimerdiol produced by hydrogenating a dimer 15acid obtained by dimerization of an unsaturated fatty acid having 11 to 22 16carbon atoms, one of the elements of the invention of claim 33 is not 17described in the prior art. Accordingly, in this case, there is no basis to 18support a prior art rejection under §103 based on Ansmann '978, with or 19without Akrongold.
- Claim 21 requires the monocarboxylic acid to be a rosin or a
 21hydrogenated rosin. Claim 21 depends indirectly from claim 33 through
 22claim 18. Apart from our discussion concerning claim 33, and assuming
 23arguendo that the Ansmann '978 "dimer diol" is the same as the Fine
 24Chemical dimerdiol, in our view the evidence will not support a finding that
 25one skilled in the art would have had a reason to use the Bernhardt rosin

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Appeal 2007-4123
Application 09/604,763
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lester in the composition of Ansmann '978. *KSR*, *supra*. Rather, the use of 2impermissible hindsight is required to find that one skilled in the art would 3have reacted rosin with Fine Chemical's dimerdiol. *Cf. In re McLaughlin*, 4443 F.2d 1392, 1395, 170 USPQ 209, 212 (CCPA 1971) (obviousness 5judgments are necessarily based on hindsight; so long as judgment takes into 6account only knowledge known in the art, there is no error). There is no 7prior art reason to react rosin with Fine Chemical's dimerdiol.

8

G. Order

- 10 Upon consideration of the appeal it is
- ORDERED that the decision of the Examiner rejecting 12claims 17 and 22-29 over the prior art is *vacated*.
- 13 FURTHER ORDERED that we have entered a new rejection of 14claims 17 and 22-29. 37 C.F.R. § 41.50(b) (2007). 35 U.S.C. § 112, second 15paragraph.
- 16 FURTHER ORDERED that the decision of the Examiner 17rejecting claims 18-21 and 33 over the prior art is *reversed*.
- 18 FURTHER ORDERED that our decision is not a final agency
- FURTHER ORDERED that within **two (2) months** from the 21date of our decision appellant may further prosecute the application on 22appeal by exercising one of the two following options:
- 23 1. Request that prosecution be reopened by submitting 24an amendment or evidence or both. 37 C.F.R. § 41.50(b)(1) (2006).
- 25 2. Request rehearing on the record presently before the 26Board. 37 C.F.R. § 41.50(b)(2) (2006).

FURTHER ORDERED that no time period for taking any 2subsequent action in connection with this appeal may be extended under 337 C.F.R. § 1.136(a)(1)(iv) (2006).

VACATED-IN-PART and REVERSED-IN-PART (New rejection under Bd. R. 41.50(b))

sd

Appeal 2007-4123 Application 09/604,763 to c (via First Class mail)

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Attorney and address

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